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**Second Semester 2019- 2020**

**Course Handout (Part-II)**

Date:28/11/2019

**In addition to Part I (General Handout for all courses appended to the Time Table), this portion gives further specific details regarding the course.**

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| **Course No*.*** | **:** | **CHE G641** |
| **Course Title** | **:** | **Reaction Engineering** |
| **Instructor-in-Charge** | **:** | **Pankaj Kumar** |
| **Other Instructors** | **:** | **SrikantaDinda, P Pradeep Reddy** |

**Course Description :**

Design of multi-phase reactors; analyses of gas-liquid and gas-liquid-solid reactions; intrinsic kinetics of catalytic reactions; residence time distribution models for micro-and macro-mixing; mathematical models for gas-liquid-solid reactors; laboratory reactors; dynamics and design of various multi-phase reactors such as trickle bed reactors, bubble column reactors, segmented-bed reactors, slurry reactors, spouted bed reactors, pulsating reactors, fluidized bed reactors, etc.; optimization of chemical reactors.

**Scope & Objective**

This course includes basic and advanced topics in Chemical Reaction Engineering. The initial part of the course includes reaction mechanisms, basic reaction kinetics (like rate, concentration, conversion and selectivity) and ideal reactor configurations. The design of catalytic and other multiphase reactors are elaborated. The characterization of non-ideal reactors and modeling of these, especially with reference to Residence Time Distribution (RTD Models) are dealt with. Thermal and Mass Transfer effects are also explained in multi-phase reactors. The main objective of this course is to give the student better understanding of Chemical Reaction Engineering of Multi Phase Non-ideal reactors encountered in Industry from the point of view of design and performance evaluation of such actual reactors.Short projects/assignments and industrial visits shall be planned to imbibe the concepts learnt better.

Learning outcomes:

After studying this course, students will be able to

* To understand, how efficiently feed can convert to product by just changing the design of reactor.
* It also gives the knowledge to what kind of reactor system is to be used for what situation.
* Student will be able to analyze the kinetic related data to find the size of reactor for a specific reaction
* Lab exposer will help to know the how to find the reaction kinetics of an unknown system

**Text Books**

**T1** Levenspiel, Octave, *Chemical Reaction Engineering*, Wiley India Pvt. Ltd., New Delhi, 3rd Edition, 2000 ( or higher edition if available!).

**T2** Fogler, H. Scott, *Elements of Chemical Reaction Engineering*, Prentice Hall of India Pvt. Ltd., New Delhi, 4th Ed., 2006 ( or higher edition if available! ).

**Reference Books**

**R1** Salmi, Tapio O., Jyri-Pekka M. and Johan P,W., *Chemical Reaction Engineering and ReactorTechnology*, CRC Press, Taylor &Francis, New York, 2011( or higher edition if available! ).

**Course Plan**

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| **Lecture No.** | **Learning Objectives** | **Topics to be covered** | **Reference** |
| 1-6 | Review of chemical kinetics and ideal reactor concepts and ideal reactor designs, conversion and sizing of ideal reactors including multiple reactors in series or parallel | Review of kinetics and ideal reactors | Chap No. 1-8 of T1  Chap No. 1-3 of T2  Chap No. 1-3 of R1 |
| 7-10 | Reaction mechanisms, elementary and non-elementary homogeneous reactions, order of reactions | Review of mechanism of reactions, order and effect on reactor design | Chap No. 1 of T1  Chap No. 7 of T2 |
| 10-12 | Laboratory reactors, collection and analysis of rate data, getting rate data | Rate models and rate expressions, Laboratory data analysis and Interpretation | Chap No. 1 of T1  Chap No. 5 of T2 |
| 13-19 | Multiple reactions – series parallel and effect on ideal reactor design | Multiple reactions | Chap No. 7-8 of T1  Chap No. 6 of T2  Chap No. 4 of R1 |
| 20-21 | Temperature and pressure effects on single and multiple reactions | Effects of temperature and pressure | Chap No. 9 of T1  Chap No. 5 of R1 |
|  |  |  |  |
| 22-24 | Catalysis and Catalytic reactors, catalyst deactivation, fouling, poisons, mitigations | Heterogeneous catalysis introduction | Chap No. 17-19 of T1  Chap No. 10 of T2 |
| 25-29 | Packed bed catalytic reactors and external and internal diffusion | Heterogeneous catalysis with mass transfer | Chap No. 19 of T1  Chap No. 10-11 of T2 |
| 30-32 | Multiphase reactors including gas solid and liquid slurry, bubble columns and fluid bed reactors, trickle bed reactors | Multiphase catalytic reactors | Chap No. 20-22 of T1  Chap No. 12 of T2 |
| 33-35 | Non-Catalytic systems, fluid-fluid, fluid-particle kinetics and reactor design | Multi-phase non-catalytic reactors | Chap No. 23-26 of T1 |
| 36-41 | Tracers, methods of obtaining Residence time Distribution(RTD), models for flow patterns | Residence time Distribution (Macro mixing) | Chap No. 11-16 of T1  Chap No. 13-14 of T2  Chap No. 6 of R1 |

**Lab experiments:**

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| **Lab Name** | **Experiment Name & Objective** |
| CRE lab | Batch Reactor: To study the order and rate constant for the reaction between KOH and ethyl acetate in a batch reactor @30 °C. |
| CRE lab | Batch Reactor: To study the order and rate constant for the reaction between KOH and ethyl acetate in a batch reactor @40 °C. |
| CRE lab | Batch Reactor: To study the order and rate constant for the reaction between KOH and ethyl acetate in a batch reactor @50 °C. |
| CRE lab | Batch Reactor: To study the order and rate constant for the reaction between KOH and butyl acetate in a batch reactor @30 °C. |
| CRE lab | Continuous Stirred Tank Reactor: To study the order and rate constant for the reaction between KOH and ethyl acetate in a CSTR @ 30 °C. |
| CRE lab | Plug Flow Reactor: To study the order and rate constant for the reaction between KOH and ethyl acetate in a PFR @ 30 °C. |
| CRE lab | RTD study using CSTR |
| CRE lab | RTD study using PFR |
| CRE lab | Analysis of gas sample |
| CRE Lab | To find conversion of a reaction using spectrophotometric method |
| CRE Lab | To find conversion of a reaction using GC analysis |
| Multiphase lab | To find adsorption capacity of an adsorbent for adsorption of gas on solid adsorbent |
| Petroleum lab | Cracking of liquid fuel |

**Evaluation scheme**

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| EC No. | Evaluation Component | Duration | Weightage% | Date, Time | Remarks |
| 1. | Mid sem | 90 min | 25 | 2/3, 11: 00 – 12:30 PM | CB |
| 2. | Compre | 180 min | 35 | 01/05 , AN | 25%CB & 10%OB |
| 3. | Assignment & / seminars (2) |  | 10 |  | OB |
| 4 | Quizzes (2) |  | 10 |  | CB |
| 5 | Lab experiments |  | 20 |  | OB |

* **Min. marks required to secure a valid grade is above 15% of total marks of all components.**
* **Chamber consultation hour**will be announced in the class.
* The **notices,** if any**,** concerning the course, will be displayed on the notice board of the Department of Chemical Engineering.
* **Make-up**will be granted for**genuine cases only**. Prior permission of IC is compulsory.

**Academic Honesty and Integrity Policy:** Academic honesty and integrity are to be maintained by all the students throughout the semester and no type of academic dishonesty is acceptable.

**Instructor-in-charge ǀ CHE G641**